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## Research Article

### DEVELOPMENT AND VALIDATION OF HPTLC METHOD FOR THE ANALYSIS OF TOLPERISONE HYDROCHLORIDE IN PHARMACEUTICAL DOSAGE FORM

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#### Keywords:

Tolperisone, HPTLC, Formulation, Estimation.

#### ABSTRACT

A simple, sensitive, rapid and precise high performance thin layer chromatographic method has been developed and validated for the estimation of Tolperisone hydrochloride in bulk and pharmaceutical dosage form. The stationary phase used was silica gel precoated aluminum plate 60F<sub>254</sub> plates. The mobile phase used was a mixture of chloroform:acetone:toluene (6:2:2, v/v/v). The detection of spots was carried out at 265 nm. The method was validated in terms of specificity, linearity, precision and accuracy. The calibration curve was found to be linear between 50-600 ng/band. The developed method was subjected for forced degradation studies like acid, alkali, peroxide and thermal stress conditions were performed as per ICH guidelines. The proposed method was suitable for routine quality control analysis of Tolperisone hydrochloride in bulk and pharmaceutical formulation.

#### INTRODUCTION

Tolperisone hydrochloride (Fig. 1) is a skeletal muscle relaxant, acts at the level of spinal cord by blocking sodium channels and calcium channels [1-2]. Chemically it is 2-methyl-1-(4-methylphenyl)-3-(1-piperidinyl)-1-propanone hydrochloride [3]. Tolperisone hydrochloride exerts its spinal reflex inhibitory action predominantly via pre-synaptic inhibition of the transmitter release from the primary afferent endings via combined action on voltage-gated sodium and calcium channels [4]. Tolperisone hydrochloride increases the blood supply to skeletal muscle and antinociceptive activity against thermal stimulation that is likely to be attributed to its local anesthetic action [5].

Literature survey revealed that few HPTLC methods [6-7] were reported for the estimation of Tolperisone hydrochloride. Hence the objective of this method is to develop and validate a simple, precise and rapid HPTLC method in accordance with ICH guidelines [8-9] for the estimation of Tolperisone hydrochloride in bulk sample and its pharmaceutical formulation.

#### EXPERIMENTAL

##### Instrumentation

To develop a high performance thin layer chromatographic method for quantitative

determination of Tolperisone hydrochloride using computerized Camag HPTLC system (Camag, Muttenz, Switzerland) consisting of a Camag 100 microlitre sample syringe (Hamilton, Bonded, Switzerland) on silica gel precoated aluminum plate 60F<sub>254</sub> plates, [20 m × 20cm width 200µm thickness; E. Merck, Darmstadt, Germany] using a Camag Linomat V (Switzerland) sample applicator. Densitometric scanning was performed using a Camag TLC scanner III in the reflectance absorbance mode at 265 nm and operated by CATS software (V 3.15, Camag). The source of radiation used was deuterium lamp emitting a continuous UV spectrum between 190 nm and 400 nm. A Camag glass twin-trough development chamber, different size pipettes, volumetric flasks, measuring cylinders, micro syringes and ruler were used.

##### Chemicals and solvents

The reference samples of Tolperisone hydrochloride was obtained as gift sample from Spectrum Pharma Research Solutions, Hyderabad, India. Commercially available tablet formulation claimed to contain 150 mg of Tolperisone hydrochloride was purchased from local market. Chloroform, acetone and toluene purchased from Merck Chemicals, Mumbai, India.

### Preparation of standard solution

Accurately weighed and transferred 50 mg of Tolperisone hydrochloride into a 100 mL clean dry volumetric flask, about 10 mL of diluent was added and volume was made upto the mark by using diluent. Further 1 mL of above stock solution was pipetted into 100 mL volumetric flask and diluted upto the mark by adding diluent. From this 1 mL of the solution was pipetted into a 100 mL volumetric flask and made upto the mark with diluent. From this serial dilutions were prepared to construct the calibration curve.

### Preparation of sample solution

Twenty tablets were accurately weighed and crushed into a fine powder in a mortar. An amount of powder equivalent to 150 mg of Tolperisone hydrochloride into 100 mL volumetric flask and about 50 mL of diluent was added. The mixture was sonicated for 20 min to dissolve and made volume upto the mark with diluent; then solution was filtered through 0.45  $\mu$ m filter paper. Further 1 mL of solution was pipetted and transferred into 100 mL volumetric flask, made volume upto the mark by using diluent. From the above stock solution, 2 mL of the solution was pipetted into a 100 mL volumetric flask and made upto volume with mobile phase to yield concentration 300 ng/band of Tolperisone hydrochloride.

### Method

TLC plates were prewashed and activation of plates was done in an oven at 50°C for 5-10 min. The chromatographic conditions maintained were precoated silica gel 60F<sub>254</sub> aluminum sheets (10x10 cm) as stationary phase, chloroform:acetone:toluene (6:2:2, v/v/v) as mobile phase. The mobile phase was used as diluent, chamber and plate saturation time of 20 min, migration distance allowed was 72 mm, keeping the slit dimension at 5x0.45 mm. A deuterium lamp provided the source of radiation. Ten microliters of standard solution of Tolperisone hydrochloride was spotted and developed at constant temperature. Wavelength was selected by scanning standard solution over 200-400 nm wavelength. Tolperisone hydrochloride show maximum absorbance at 265 nm in reflectance mode with Camag TLC Scanner 3 using WinCATS software.

### Calibration curve

Aliquot of 50, 100, 200, 300, 400, 500 and 600 ng/band of standard solution of Tolperisone hydrochloride was applied on the TLC plate. TLC plate was dried, developed and analyzed photometrically as described earlier. The standard

calibration curve was plotted using regression analysis with Microsoft excel.

### Validation of the method

The developed method was validated in terms of specificity, linearity, system precision, method precision and intermediate precision, accuracy, limit of detection and limit of quantitation.

### STABILITY STUDIES

Forced degradation studies were conducted indicating the stability of proposed method by subjecting the sample for acid degradation, alkali degradation, peroxide degradation and thermal degradation.

### Control sample

Twenty tablets were accurately weighed and crushed in to a fine powder in a mortar. An amount of powder equivalent to 150 mg of Tolperisone hydrochloride into 100 mL volumetric flask and 50 mL of diluent was added to it. The mixture was sonicated for 20 min to dissolve and made volume up to the mark with diluent; the solution was filtered through 0.45  $\mu$ m filter paper. Further 1 mL of solution was transferred to 100 mL volumetric flask, made volume upto the mark by using diluent. From the above stock solution, 2 mL of the solution was pipetted into a 100 mL volumetric flask and made upto the mark with diluent.

### Acid degradation studies

Twenty tablets were weighed and finely powdered. An accurately weighed portion of powder sample equivalent to 150 mg of Tolperisone hydrochloride was transferred to a 100 mL volumetric flask containing 50 mL of the diluent. The contents of the flask were sonicated for about 30 min for complete solubility of the drug with intermittent shaking at controlled temperature and the volume was made up to the mark by using diluent. From this 1 mL of the solution was pipetted into 100 mL volumetric flask and made volume upto the mark with diluent. Then 5 mL of 0.1N acid (Hydrochloric acid) was added, refluxed for 8 hours at 60°C, then cooled to room temperature, neutralized with 0.1N base (Sodium hydroxide) and diluted to volume with diluent. Then this mixture was filtered through 0.45  $\mu$ m membrane filter. 2 mL of the above filtered sample solution was pipetted into a 100 mL volumetric flask and diluted to volume with diluent.

### Alkali degradation studies

Twenty tablets were weighed and finely powdered. An accurately weighed portion of powder sample equivalent to 150 mg of Tolperisone hydrochloride was transferred to a 100 mL volumetric flask containing 50 mL of the diluent. The contents of the

flask were sonicated for about 30 min for complete solubility of the drug with intermittent shaking at controlled temperature and the volume was made upto the mark by using diluent. From this 1 mL of the solution was pipetted into 100 mL volumetric flask and made volume upto the mark with diluent. Then 5 mL of 0.1N base (Sodium hydroxide) was added, refluxed for 8 hours at 60°C, then cooled to room temperature, neutralized with 0.1N acid (Hydrochloric acid) and diluted to volume with diluent. Then this mixture was filtered through 0.45  $\mu$  membrane filter. Pipetted 2.0 mL of the above filtered sample solution into a 100 mL volumetric flask and diluted to volume with diluent.

#### Peroxide degradation studies

Twenty tablets were weighed and finely powdered. An accurately weighed portion of powder sample equivalent to 150 mg of Tolperisone hydrochloride was transferred to a 100 mL volumetric flask containing 50 mL of the diluent. The contents of the flask were sonicated for about 30 min for complete solubility of the drug with intermittent shaking at controlled temperature volume was made upto the mark by using diluent. From this 1 mL of the solution was transferred into 100 mL volumetric

flask and made volume upto the mark with diluent. Then 2 mL of 3 % hydrogen peroxide was added, refluxed for 8 hours at 60°C, then cooled to room temperature and diluted to volume with diluent. Then this mixture was filtered through 0.45  $\mu$  membrane filter. 2.0 mL of the above filtered sample solution was pipetted into a 100 mL volumetric flask and diluted to volume with diluent.

#### Thermal degradation studies

Twenty tablets were weighed and finely powdered. The powder is exposed to heat at 105°C for about 8 hours. An accurately weighed portion of powder sample equivalent to 150 mg of Tolperisone hydrochloride was transferred to a 100 mL volumetric flask containing 50 mL of the diluent. The contents of the flask were sonicated for about 30 min for complete solubility of the drug with intermittent shaking at controlled temperature and then cooled the solution to room temperature and volume made up with further quantity of diluent. Then this mixture was filtered through 0.45  $\mu$  membrane filter. 3.3 mL of the above filtered sample solution was pipetted into a 100 mL volumetric flask and diluted to volume with diluent.

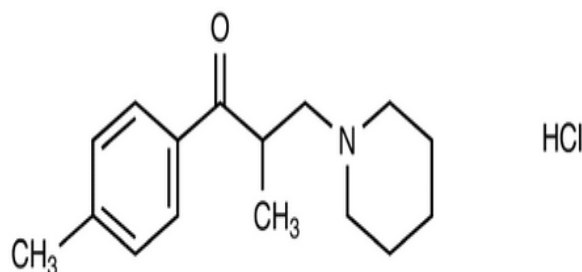


Fig. 1: Chemical structure of Tolperisone hydrochloride

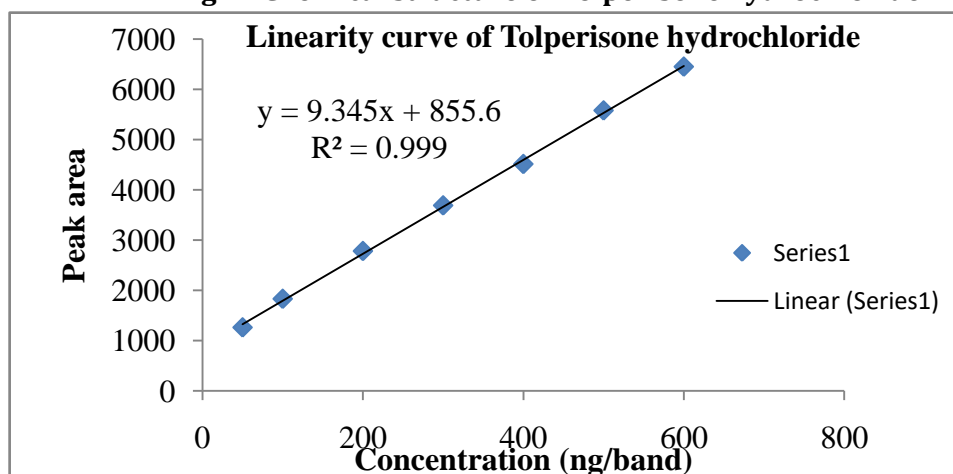


Fig. 2: Calibraton curve of Tolperisone hydrochloride

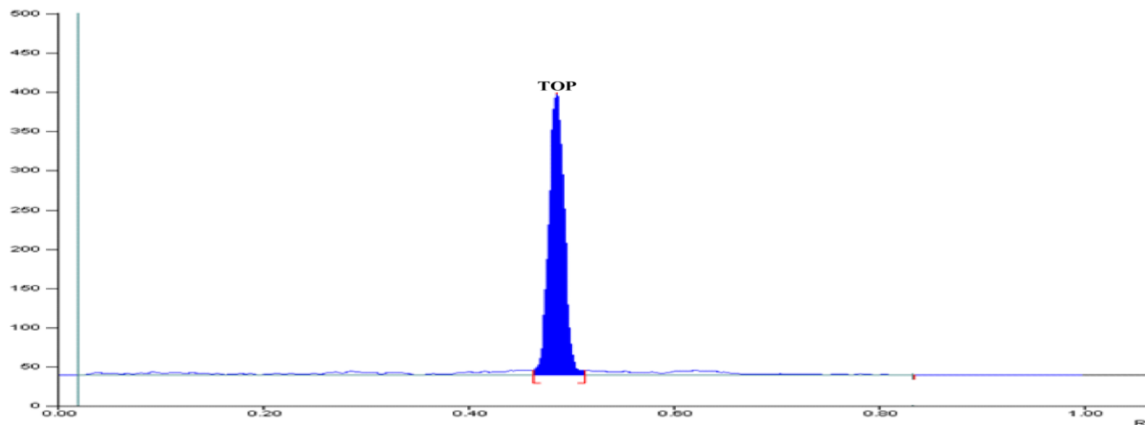


Fig. 3: Typical densitogram of Tolperisone hydrochloride

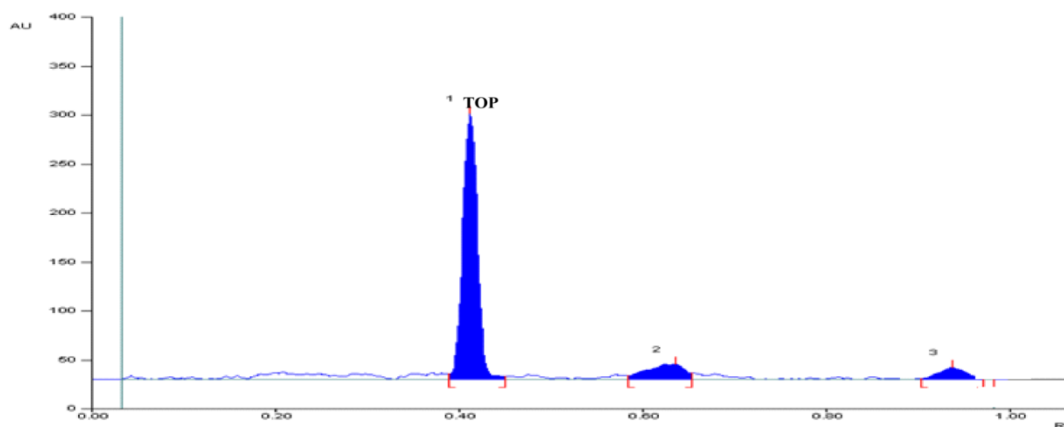


Fig. 4: Densitogram of acid degradation showing Tolperisone hydrochloride

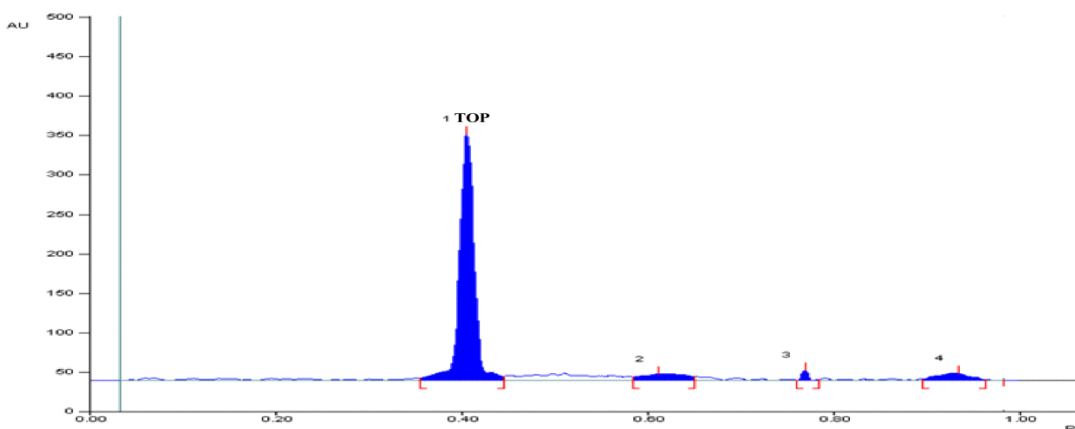


Fig. 5: Densitogram of alkali degradation showing Tolperisone hydrochloride

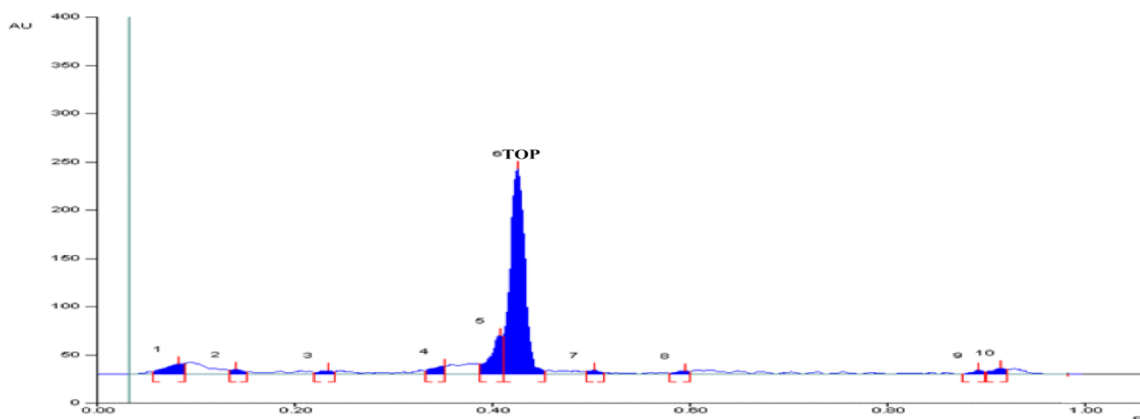


Fig. 6: Densitogram of peroxide degradation showing Tolperisone hydrochloride

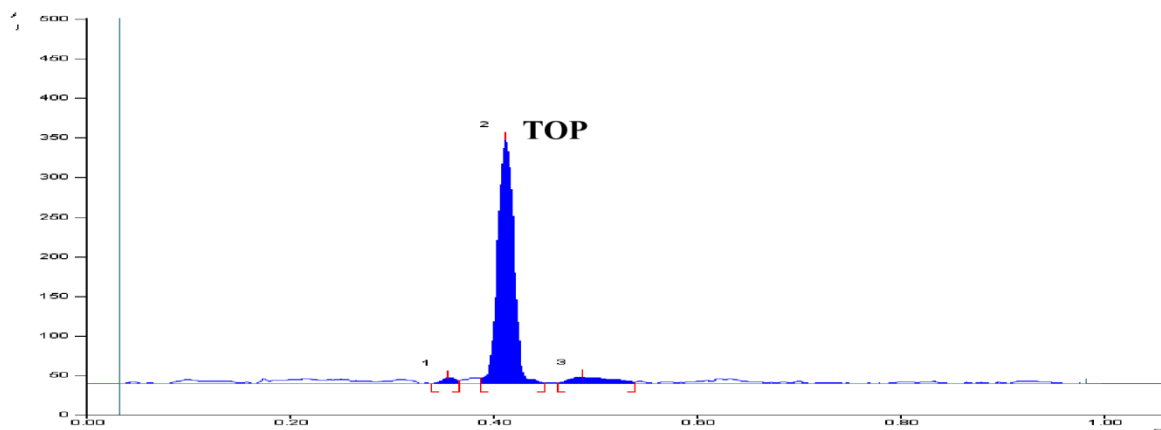


Fig. 7: Densitogram of thermal degradation showing Tolperisone hydrochloride

Table 1: Optimized chromatographic conditions of Tolperisone hydrochloride

Parameter	Condition
Mobile phase	Chloroform:acetone:toluene (6:2:2, v/v/v)
Diluent	Mobile phase
Stationary phase	Silica gel pre-coated aluminum plate 60F <sub>254</sub> plates
Saturation time	20 min
Wave length	265 nm
Sample volume	10 $\mu$ L
R <sub>f</sub>	0.41

Table 2: Linearity study of Tolperisone hydrochloride

Concentration (ng/band)	Mean peak area
50	1257.85
100	1826.93
200	2778.24
300	3685.82
400	4509.86
500	5575.69
600	6447.2
Slope	9.345
Intercept	855.6
Correlation Coefficient	0.999

Table 3: Validation parameters of Tolperisone hydrochloride

Parameters	Values
Linearity range (ng/band)	50-600
Limit of detection (LOD)	10 ng/band
Limit of quantitation (LOQ)	30 ng/band
Precision (% RSD)	
System precision (n=6)	1.3
Method precision (n=6)	0.78
Intermediate precision (n=6)	0.33

**Table 4: Recovery results of Tolperisone hydrochloride**

Level	Concentration added (ng/band)	Concentration found (ng/band)	% Recovery	Mean recovery
50%	150	148.9	99.26%	99.64%
100%	300	299.44	99.81%	
150%	450	449.41	99.87%	

\*Each value is a mean ± standard deviation of three determinations.

**Table 5: Degradation studies of Tolperisone hydrochloride**

Stress conditions	Degradation time	% Assay	% Degradation
Control	8 hours	100.0	-
Acidic	8 hours	97.6	2.4
Alkali	8 hours	95.23	4.77
Peroxide	8 hours	98.4	1.6
Thermal	8 hours	98.2	1.8

## RESULTS AND DISCUSSION

For HPTLC, silica gel precoated aluminum plates 60F<sub>254</sub> was chosen as the stationary phase for the separation and determination of Tolperisone hydrochloride. Mixture of commonly used solvents like butane:ethyl acetate, chloroform:butanol:toluene combinations were tested as mobile phase. The choice of the optimum composition was based on the chromatographic response factor, a good peak shape with minimum tailing. A mixture of chloroform:acetone:toluene (6:2:2, v/v/v) was proved to be the most suitable of all the combinations since the densitogram peak obtained was well defined peaks are measured at 265 nm, better resolved. The results of optimized chromatographic conditions of Tolperisone hydrochloride were shown in Table 1.

The specificity of method was performed by comparing the densitograms of blank, standard and sample. The absence of additional peaks in the densitogram indicates non-interference of the commonly used excipients in the tablets and hence the method was specific. The linearity was found 50-600 ng/band of Tolperisone hydrochloride. The regression equation of the linearity curve between concentrations of Tolperisone hydrochloride over its peak areas were found to be  $Y=9.345X+855.6$  (where Y is the peak area and X is the concentration of Tolperisone hydrochloride in ng/band) and the correlation coefficient was 0.999. The linearity curve of Tolperisone hydrochloride was shown in Fig. 2. The linearity results were furnished in Table 2.

To study the system precision, six replicate standard solutions of Tolperisone hydrochloride was injected. The %RSD was calculated and it was found to be 1.3 of Tolperisone hydrochloride, which are well within the acceptable criteria of not more than 2.0. The method precision study was carried out on six preparations from the same tablet samples of Tolperisone hydrochloride percent amount of both were calculated. The %RSD of the assay result of six preparations in method precision study was found to be 0.78. The intermediate precision study was carried out by different analysts; from the same tablet of Tolperisone hydrochloride in the percent amount of Tolperisone hydrochloride was calculated. The %RSD of the assay result of six preparations in intermediate precision study was 0.33. The lower %RSD values reveals that the method quite precise.

The limit of detection and limit of quantitation for Tolperisone hydrochloride was found to be 10 ng/band and 30 ng/band. The lowest values of LOD and LOQ as obtained by the proposed method indicate that the method was sensitive. The method validation parameters were shown in Table 3.

The accuracy of the method was determined by recovery studies. The percent recoveries of the drug solutions were studied at three different concentration levels. The percent recovery and %RSD at each level was calculated. Satisfactory recoveries ranging from 99.26 to 99.87% of Tolperisone hydrochloride was obtained by the proposed method. The percent individual recovery and the %mean recovery at each level were within

the acceptable limits. This indicates that the method was accurate. The recovery studies results are furnished in Table 4. The  $R_f$  value of the Tolperisone hydrochloride was found to be 0.41. The densitogram of Tolperisone hydrochloride was shown in Fig. 3.

HPTLC studies of Tolperisone hydrochloride under different stress conditions indicated the following degradation behavior. In acid degradation, the % degradation was 2.4%. In alkali degradation, the % degradation was 4.77%. In peroxide degradation, the % degradation was 1.6%. In thermal degradation, the % degradation was 1.8%. The results of degradation studies are given in Table 5. The typical chromatograms of degradation behavior of Tolperisone hydrochloride in different stress conditions are shown in Fig. 4 to Fig. 7.

### CONCLUSION

The developed HPTLC method for the estimation of Tolperisone hydrochloride is simple, specific, precise, accurate and reproducible. The amounts found in formulations are well agreed with label claim. The proposed method can be applied to routine analysis in quality control laboratories for the estimation of Tolperisone hydrochloride in bulk and in pharmaceutical formulation.

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